

Psilocin Analogs. 1. Synthesis of 3-[2-(Dialkylamino)ethyl]-
and 3-[2-(Cycloalkylamino)ethyl]indol-4-ols

David B. Repke*

148-B West Dana Street, Mountain View, California 94041

Wilfred J. Ferguson

Synchemex Laboratories, 2860 Detroit Ave., Cleveland, Ohio 44113

Dallas K. Bates

Department of Chemistry and Chemical Engineering,
Michigan Technological University, Houghton, Michigan 49931

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The synthesis of four dialkyl and three cycloalkyl analogs of psilocin (**4**, R = CH₃), a hallucinogenic principle found in certain fungi, is described. The synthetic route involves four transformations starting with 6,7-dihydroindol-4(5H)one (**1**).

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The rediscovery (1,2) of the ceremonial use of certain agarics in Mexico led Hofmann and collaborators to a phytochemical investigation of the species involved (3). Animal and human testing (4) of various fractions from extracts of cultivated *Psilocybe mexicana* Heim resulted in the isolation of two psychoactive principles. Structural studies and confirmation by synthesis (5-7) showed these compounds to be 3-[2-(dimethylamino)ethyl]indol-4-ol (psilocin, **4**, R = CH₃) and its corresponding dihydrogen-phosphate ester (psilocybin). These two substances have subsequently been detected in various species of the genera *Psilocybe* (Fr.) Quél., *Panaeolus* (Fr.) Quél. and *Conocybe* Fayod (8).

The hallucinogenic effects of psilocin and psilocybin have been well documented (9-12). Delay, Pichot and Lempérière (13) have commented on the possible psychotherapeutic role of these substances, although the length and intensity of the experience might preclude their use in all but the most special circumstances. Structural modification using psilocin as a template might lead to compounds having mood-enhancing qualities instead of hallucinogenic effects. Such an approach has been utilized by Shulgin, Sargent and Naranjo (14) in an extensive program of the structural modification of mescaline. Some success in the use of these analogs as adjuncts to psychotherapy has been reported by Naranjo (15).

Troxler, Seemann and Hofmann (16) synthesized a number of analogs of **4** with variations in the position of the oxygen substituent as well as modifications of the

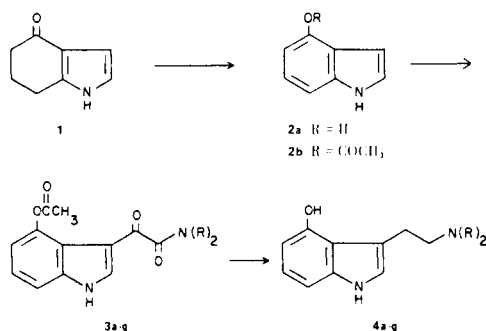
ethylamino side chain. The diethyl analog of psilocin (Sandoz CZ-74) reached clinical trial (17) and was found to be as active as psilocin but with a shorter duration.

Psilocin has been synthesized by a number of routes (7, 18-21). Hofmann's original synthesis has recently been repeated by Ono, Shimamine and Takahashi (22).

An attractive entry into 4-hydroxyindole chemistry is via dehydrogenation (23) of 6,7-dihydroindol-4(5H)one, **1**. Compound **1** has been prepared from 1,3-cyclohexanedione (24) and by a three step synthesis from pyrrole (18). Extensive exploitation of this route has not occurred although a notable exception is the work of Remers, Roth, Gibs and Weiss (25,26).

Compound **1** was conveniently aromatized with 10% palladium on carbon in refluxing *p*-cymene. The 4-hydroxyindole thus obtained was acetylated to provide crystalline 4-acetoxyindole (**2b**). Compound **2b** was then

Scheme 1



converted to the glyoxylamides (3) and thence to the amines (4) by the method of Speeter and Anthony (27). (Scheme I). The entire sequence results in the production of modest yields of the amines (4) through crystalline intermediates without the need for chromatographic purification at any step.

Early experiments on the isolation of the amines (4) from the reduction mixtures demonstrated their lability in the presence of impurities. Filtration of hydrolyzed reduction complexes led to rapid decomposition of the amines with formation of deep blue pigmented solutions. A similar reaction occurred during attempted chromatography of the crude amines on silicic acid. However, it

was found that work-up of the reduction mixtures in an inert atmosphere minimized pigment formation. The amines were then purified by either distillation or sublimation *in vacuo*. The pure amines were found to be stable when stored in an anhydrous atmosphere at -15° . The diisopropyl analog (4d) could not be induced to crystallize as the free base but was converted to the hydrochloride salt by careful addition of 0.1N hydrochloric acid to a solution of the amine in methanol.

The proton magnetic resonance spectra appear to be characteristic for this type of compound (28). The spectra of 3a-3d exhibit a doubling up of signals for the amide alkyl substituents due to the partial double bond character

Table I

Physical Data for the Amides 3

Compound	Yield	R	M.p. (solvent)	Formula	Analyses
3a	74%	ethyl	150-151 (ether)	$C_{16}H_{18}N_2O_4$ (302.34)	Calcd: C, 63.56; H, 6.00; N, 9.27 Found: C, 63.57; H, 5.87; N, 8.95
3b	78%	n-propyl	130-131 (ether/ cyclohexane)	$C_{18}H_{22}N_2O_4$ (330.40)	Calcd: C, 65.44; H, 6.71; N, 8.48 Found: C, 65.10; H, 6.79; N, 8.35
3c	77%	n-butyl	123-125 (cyclo- hexane/hexane)	$C_{20}H_{26}N_2O_4$ (358.45)	Calcd: C, 67.02; H, 7.31; N, 7.82 Found: C, 66.77; H, 7.62; N, 7.64
3d	35%	iso-propyl	204-206 (ethyl acetate/hexane)	$C_{18}H_{22}N_2O_4$ (330.40)	Calcd: C, 65.44; H, 6.71; N, 8.48 Found: C, 65.08; H, 6.76; N, 8.26
3e	55%	pyrrolidyl	174-176 (chloro- form/hexane)	$C_{16}H_{16}N_2O_4$ (300.33)	Calcd: C, 63.98; H, 5.38; N, 9.32 Found: C, 63.67; H, 5.58; N, 9.02
3f	60%	piperidyl	177-178 (chloro- form/hexane)	$C_{17}H_{18}N_2O_4$ (314.35)	Calcd: C, 64.96; H, 5.77; N, 8.91 Found: C, 64.89; H, 5.59; N, 8.62
3g	80%	morpholidyl	190-191 (ethyl acetate/hexane)	$C_{16}H_{16}N_2O_5$ (316.33)	Calcd: C, 60.75; H, 5.10; N, 8.86 Found: C, 60.80; H, 5.10; N, 8.88

Table II

Physical Data for the Amines 4

Compound	Yield	R	M.p. (solvent)	Formula	Analyses
4a	50%	ethyl	103-104 (ethyl acetate/hexane) (a)	$C_{14}H_{20}N_2O$ (232.33)	Calcd: C, 72.38; H, 8.68; N, 12.06 Found: C, 72.46; H, 8.73; N, 11.91
4b	51%	n-propyl	96-97 (ethyl acetate/hexane)	$C_{16}H_{24}N_2O$ (260.39)	Calcd: C, 73.80; H, 9.29; N, 10.76 Found: C, 73.81; H, 9.27; N, 10.75
4c	35%	n-butyl	74-75 (ethyl acetate/hexane)	$C_{18}H_{28}N_2O$ (288.44)	Calcd: C, 74.95; H, 9.78; N, 9.71 Found: C, 74.88; H, 9.86; N, 9.66
4d	47%	iso-propyl	263 dec. (methanol/ ether) (b)	$C_{16}H_{25}N_2ClO$ (296.86)	Calcd: C, 64.74; H, 8.49; N, 9.44 Found: C, 64.56; H, 8.48; N, 9.29
4e	50%	pyrrolidyl	193-195 (ethyl acetate/hexane)	$C_{14}H_{18}N_2O$ (230.31)	Calcd: C, 73.01; H, 7.88; N, 12.17 Found: C, 72.98; H, 7.92; N, 12.05
4f	42%	piperidyl	180-181 (ethyl acetate/hexane) (c)	$C_{15}H_{20}N_2O$ (244.35)	Calcd: C, 73.73; H, 8.25; N, 11.47 Found: C, 73.67; H, 8.56; N, 11.38
4g	46%	morpholidyl	177-178 (ethyl acetate/hexane)	$C_{14}H_{18}N_2O_2$ (246.31)	Calcd: C, 68.27; H, 7.36; N, 11.38 Found: C, 68.36; H, 7.44; N, 11.30

(a) Reference 16 reports m.p. 104-106°. (b) Characterized as the hydrochloride salt. (c) Reference 16 reports m.p. 182-183°.

of the C-N bond (29). The *ABC* spin system (30) of the aromatic C₅H, C₆H and C₇H signals is not amenable to first-order analysis. However, approximate parameters were obtained in certain well-resolved cases, e.g., 4f.

EXPERIMENTAL

Proton magnetic resonance spectra were obtained with a Varian Associates T-60 spectrometer and are reported in ppm δ downfield from an internal standard of tetramethylsilane. Elemental analyses

Table III

Proton Magnetic Resonance Parameters for the Amides 3

Compound	Solvent	Chemical Shift, ppm δ
3a	Deuteriochloroform	7.53 (br s, 1H, N ₁ H), 7.10 (s, 1H, C ₂ H), 7.05 (m, 3H, C ₅ H, C ₆ H, C ₇ H), 3.36 (m, 4H, J = 8 Hz, ethyl CH ₂), 2.45 (s, 3H, acetate CH ₃), 1.20 (t, 3H, J = 8 Hz, ethyl CH ₃), 1.06 (t, 3H, J = 8 Hz, ethyl CH ₃)
3b	Deuteriochloroform	2.44 (br s, 1H, N ₁ H), 6.90 (m, 4H, C ₂ H, C ₅ H, C ₆ H, C ₇ H), 3.30 (m, 4H, propyl CH ₂), 2.44 (s, 3H, acetate CH ₃), 1.62 (m, 4H, propyl CH ₂), 0.86 (m, 6H, propyl CH ₃)
3c	Deuteriochloroform	7.51 (br s, 1H, N ₁ H), 6.97 (m, 4H, C ₂ H, C ₅ H, C ₆ H, C ₇ H), 3.30 (m, 4H, butyl CH ₂), 2.45 (s, 3H, acetate CH ₃), 1.54 (m, 8H, butyl CH ₂), 0.97 (m, 6H, butyl CH ₃)
3d	Deuteriochloroform plus dimethylsulfoxide-d ₆ , 1:1	7.86 (br s, 1H, N ₁ H), 7.20 (m, 4H, C ₂ H, C ₅ H, C ₆ H, C ₇ H), 3.67 (m, 2H, isopropyl CH), 2.42 (s, 3H, acetate CH ₃), 1.56 (d, 6H, isopropyl CH ₃), 1.18 (d, 6H, isopropyl CH ₃)
3e	Deuteriochloroform	8.05 (br s, 1H, N ₁ H), 7.11 (m, 4H, C ₂ H, C ₅ H, C ₆ H, C ₇ H), 3.50 (m, 4H, pyrrolidyl α -CH ₂), 2.42 (s, 3H, acetate CH ₃), 1.88 (m, 4H, pyrrolidyl β -CH ₂)
3f	Deuteriochloroform plus dimethylsulfoxide-d ₆ , 10:1	7.91 (br s, 1H, N ₁ H), 7.20 (m, 4H, C ₂ H, C ₅ H, C ₆ H, C ₇ H), 3.46 (m, 4H, piperidyl CH ₂), 2.46 (s, 3H, acetate CH ₃), 1.67 (br s, 6H, piperidyl CH ₂)
3g	Deuteriochloroform	7.98 (br s, 1H, N ₁ H), 7.20 (m, 4H, C ₂ H, C ₅ H, C ₆ H, C ₇ H), 3.58 (m, 4H, morpholidyl O-CH ₂), 2.86 (br s, 4H, morpholidyl N-CH ₂), 2.41 (s, 3H, acetate CH ₃)

Table IV

Proton Magnetic Resonance Parameters for the Amines 4

Compound	Solvent	Chemical Shift, ppm δ
4a	Deuteriochloroform	8.05 (br s, 1H, N ₁ H, exchange with deuterium oxide), 6.81 (m, 2H, C ₆ H, C ₇ H), 6.77 (s, sharpening on addition of deuterium oxide, C ₂ H), 6.52 (dd, 1H, C ₅ H, J _{5,6} = 8 Hz, J _{5,7} = 2 Hz), 2.80 (m, 4H, α -CH ₂ , β -CH ₂), 2.56 (dd, 4H, J = 7 Hz, ethyl CH ₂), 0.97 (t, 6H, J = 7 Hz, ethyl CH ₃)
4b	Deuteriochloroform	7.92 (br s, 1H, N ₁ H, exchange with deuterium oxide), 7.05 (t, 1H, C ₆ H, J _{5,6} = J _{6,7} = 8 Hz), 6.85 (dd, 1H, C ₇ H, J _{5,7} = 2 Hz, J _{6,7} = 8 Hz), 6.83 (s, 1H, C ₂ H, sharpening on addition of deuterium oxide), 6.53 (dd, 1H, C ₅ H, J _{5,6} = 8 Hz, J _{5,7} = 2 Hz), 2.86 (m, 4H, α -CH ₂ , β -CH ₂), 2.52 (m, 4H, propyl CH ₂), 1.53 (m, 4H, propyl CH ₂), 0.83 (t, 6H, propyl CH ₃ , J = 6 Hz)
4c	Deuteriochloroform	7.97 (br s, 1H, N ₁ H, exchange with deuterium oxide), 6.93 (m, 2H, C ₆ H, C ₇ H), 6.73 (s, 1H, C ₂ H, sharpening on addition of deuterium oxide), 6.52 (dd, 1H, C ₅ H, J _{5,6} = 7 Hz, J _{5,7} = 2 Hz), 2.83 (m, 4H, α -CH ₂ , β -CH ₂), 2.46 (m, 4H, butyl CH ₂), 1.30 (m, 8H, butyl CH ₂), 0.77 (m, 6H, butyl CH ₃)
4d	Deuterium oxide plus dimethylsulfoxide-d ₆ , 1:1	7.00 (m, 4H, C ₂ H, C ₅ H, C ₆ H, C ₇ H), 3.55 (m, 2H, isopropyl CH, J = 7 Hz), 3.16 (br s, 4H, α -CH ₂ , β -CH ₂), 1.15 (d, 12H, isopropyl CH ₃ , J = 7 Hz)
4e	Deuteriochloroform plus acetone-d ₆ , 1:1	6.95 (m, 3H, C ₂ H, C ₆ H, C ₇ H), 6.53 (dd, 1H, C ₅ H, J = 7 Hz), 2.98 (br s, 4H, α -CH ₂ , β -CH ₂), 2.60 (m, 4H, pyrrolidyl CH ₂), 1.97 (m, 4H, pyrrolidyl CH ₂)
4f	Deuteriochloroform	7.92 (br s, 1H, N ₁ H, exchange with deuterium oxide), 7.03 (t, 1H, C ₆ H, J _{5,6} = J _{6,7} = 7 Hz), 6.83 (dd, 1H, C ₇ H, J _{5,7} = 2 Hz, J _{5,6} = 7 Hz), 6.81 (br s, sharpening on addition of deuterium oxide, 1H, C ₂ H), 6.55 (dd, 1H, C ₅ H, J _{5,6} = 7 Hz, J _{5,7} = 2 Hz), 2.92 (m, 2H, β -CH ₂), 2.60 (m, 2H, α -CH ₂), 2.53 (m, 4H, piperidyl CH ₂)
4g	Deuteriochloroform	7.97 (br s, 1H, N ₁ H, exchange with deuterium oxide), 6.96 (m, 2H, C ₆ H, C ₇ H), 6.85 (s, 1H, C ₂ H, sharpening on addition of deuterium oxide), 6.62 (dd, 1H, C ₅ H, J _{5,6} = 7 Hz, J _{5,7} = 2 Hz), 3.88 (m, 4H, morpholidyl CH ₂), 3.05 (m, 2H, β -CH ₂), 2.73 (m, 4H, α -CH ₂ and morpholidyl CH ₂)

were performed by Galbraith Laboratories, Inc., Knoxville, Tennessee, and by the Microanalytical Laboratory, Department of Chemistry and Chemical Engineering, Michigan Technological University. Melting points are corrected. Reactions were monitored by thinlayer chromatography on 250 μ layers of silica gel GF on glass plates.

4-Hydroxyindole (2a).

Under a nitrogen atmosphere 10 g. (74 mmoles) of 6,7-dihydroindol-4(5H)one (24) and 2.5 g. of 10% palladium on carbon were stirred and refluxed in 250 ml. of *p*-cymene for 24 hours. The reaction was cooled to 40° and the catalyst removed by filtration through Celite. The filtered solid was washed with 100 ml. of methanol and the combined filtrate and washings were concentrated under reduced pressure. The residue was crystallized from cyclohexane, 7.6 g. (77%), m.p. 96-97° (Literature (18) m.p. 97°).

4-Acetoxyindole (2b).

Compound 2a (7.6 g., 57 mmoles) was dissolved in 200 ml. of pyridine and 150 ml. of acetic anhydride added. The reaction was stored at room temperature for 18 hours. The solvent was distilled *in vacuo* and the residue co-distilled with two 100 ml. portions of toluene. The solid residue was recrystallized from cyclohexane to give 8.07 g. (81%), m.p. 98-100° (Literature (18) m.p. 99°); nmr (deuteriochloroform): ppm δ 8.30 (br s, 1H, N₁H, exchange with deuterium oxide), 7.03 (m, 4H, C₂H, C₅H, C₆H, C₇H), 6.42 (m, 1H, C₃H, collapsing to d on addition of deuterium oxide), 2.41 (s, 3H, acetate CH₃).

4-(Acetoxy)-*N,N*-dialkyl- and -*N,N*-Cycloalkylindole-3-glyoxylamides (3a-g).

To a stirred and cooled (0°) solution of 0.5 ml. (5.9 mmole) of oxalyl chloride in 3.0 ml. of dry ethyl ether was added dropwise a solution of compound 2b (500 mg., 2.85 mmoles) in 4.0 ml. of ether. The reaction was stirred at 0° for 5 hours and then a solution of 40% dialkyl- (or cycloalkyl-) amine in ether was added dropwise until the pH was 8-9. The reaction was diluted with chloroform (100 ml.) and shaken with 30 ml. of 5% sodium bisulfate solution, 30 ml. of saturated sodium bicarbonate solution and 30 ml. of saturated sodium chloride solution. After drying, (magnesium sulfate) the organic solution was concentrated under reduced pressure. The residue was recrystallized from the appropriate solvent.

3-[2(Dialkyl- (or Cycloalkyl-)amino)ethyl]indol-4-ols (4a-g).

To a stirred suspension of lithium aluminum hydride (13 mmoles) in 10 ml. of tetrahydrofuran (previously distilled over sodium) under a nitrogen atmosphere at room temperature was added dropwise a solution of the above amide (3a-g) (2.0 mmoles) in 5-10 ml. of tetrahydrofuran at such a rate as to maintain a gentle reflux. After the addition (7-10 minutes) the reaction was refluxed for an additional 15 minutes. The reaction was cooled to 40° and the excess reagent and complex decomposed by dropwise addition of 1.0 ml. of ethyl acetate and then 2-3 ml. of water. The mixture was filtered in an anhydrous nitrogen atmosphere and the filtrate concentrated under reduced pressure. The residue was either distilled (4a-d) or sublimed (4e-g) *in vacuo* and then recrystallized.

3-[2(Diisopropylamino)ethyl]indol-4-ol Hydrochloride (4d).

The oily free base of 4d obtained by distillation of the crude amine, (0.5 mmole) was dissolved in 2.0 ml. of methanol and one equivalent of 0.1N hydrochloric acid was added. The solution was concentrated to dryness in a stream of nitrogen. All traces of moisture were removed by evacuation on the oil pump. The solid residue was recrystallized from methanol/ether.

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